

# Asymmetric [2+2+1] cyclopentannulation of olefins. Ring expansion of 2-*N*-methyl-*N*-tosyl-cyclobutanone

Florence Mahuteau-Betzer<sup>a</sup> and Léon Ghosez<sup>a,b,\*</sup>

<sup>a</sup>Department of Chemistry, University of Louvain, 1 Place Louis Pasteur, 1348 Louvain-la-Neuve, Belgium

<sup>b</sup>European Institute of Chemistry and Biology, IECB-ENSCP, 16 Avenue Pey-Berland, 33607 Pessac, France

Received 13 April 2002; accepted 15 May 2002

**Abstract**— $\alpha$ -*N*-Methyl-*N*-tosyl cyclobutanones **2** which had been previously prepared in good yields and high enantiomeric excesses from olefins and chiral keteniminium salts have been converted into the corresponding oxiranes **3** by reaction with dimethylsulfonium methylid. The stereochemistry of this reaction was found to be dependent on several factors which have been analyzed. Treatment of these oxiranes with a stoichiometric amount of lithium iodide in refluxing tetrahydrofuran gave excellent yields of monocyclic or fused cyclopentenones **4** resulting from a  $\beta$ -elimination of *N*-methyl-*N*-tosylamide from a primarily formed cyclopentanone. The ring-expansion was totally selective but for oxiranes attached to a bicyclo[4.2.0]octanone system. In all cases, the enantiomeric purities of the starting cyclobutanones were preserved throughout the sequence which thus represents a useful [2+2+1] strategy for the cyclopentannulation of olefins. © 2002 Elsevier Science Ltd. All rights reserved.

## 1. Introduction

Cyclobutanones and cyclobutenones are very useful intermediates which have found applications in total syntheses of complex molecular structures.<sup>1</sup> As a consequence of their high angle strain, the electrophilicity of the carbonyl group is enhanced and carbon-carbon bonds are weakened offering possibilities for mild and chemoselective cleavage of the four-membered ring. The cheapest and most general method to prepare these four-membered ring ketones involves a [2+2] cycloaddition of in situ generated ketenes or keteniminium salts.<sup>2</sup> The cycloaddition of ketenes with olefins bearing a chiral substituent<sup>3</sup> or the use of keteniminium salts derived from amides bearing a chiral auxiliary<sup>4</sup> have opened a practical route towards enantiomerically pure cyclobutanones and further extended their utility in synthesis.

Early studies of our laboratory have shown that the asymmetric cycloadditions of keteniminium salt **1** to 1,2-disubstituted olefins followed by a regioselective Baeyer–Villiger oxidation of the resulting cyclobutanone was a unique method for the enantioselective vicinal acylation of an olefin (Scheme 1).<sup>5</sup> More recently, we have examined the transformation of these enantiomerically enriched cyclobutanones into cyclopentenones.<sup>6</sup> We report herein the full details of these studies.

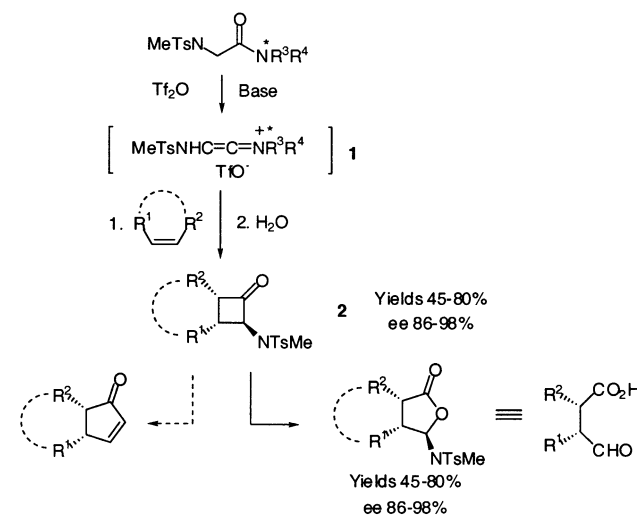
**Keywords:** cyclopentannulation; rearrangement; cyclobutanone; cyclopentenone.

\* Corresponding author. Tel.: +32-10-47-27-41; fax: +32-10-47-29-44; e-mail: ghosez@chim.ucl.ac.be

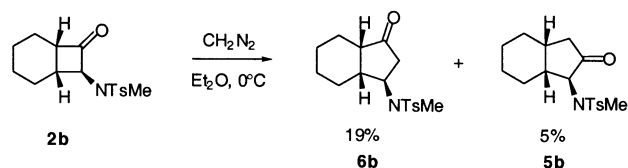
## 2. Results and discussion

### 2.1. Preliminary results

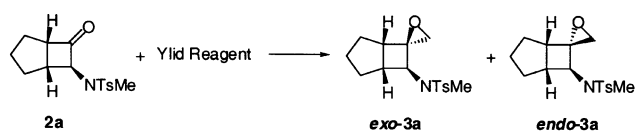
The ring enlargement of cyclobutanones with diazomethane is well documented in spite of its limited preparative value.<sup>1c</sup> The regioselectivity of the reaction is subtly dependent on substitution at both  $\alpha$  and  $\beta$  carbon atoms.<sup>7</sup> From our previous results on Baeyer–Villiger oxidation of **2**, we had anticipated that the reaction of **2b** with diazomethane would predominantly lead to regioisomer **6b**. This was indeed the case (Scheme 2). However bicyclo[4.2.0]octanone reacted sluggishly and yields were



Scheme 1.



Scheme 2.



$\text{Me}_3\text{S}^+\text{BF}_4^- + n\text{-BuLi}$	69% (1:1)
$\text{Me}_3\text{S}^+\text{Br}^- + t\text{-BuOK}$	0%
$\text{Me}_3\text{S}^+\text{I}^- + \text{NaH-DMSO}$	0%
$\text{Me}_3\text{S}^+\text{O}^- + \text{NaH-DMSO}$	0%

Scheme 3.

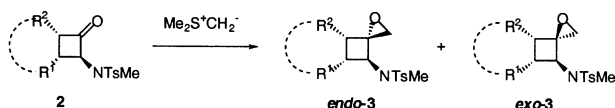
low. We therefore decided to study a sequence whereby we first generate an epoxide and then effect the rearrangement of this oxaspiropentane following the procedure developed by the groups of Lriverend and Trost.<sup>8</sup>

## 2.2. Synthesis of oxiranes

A preliminary study of the reaction of **2a** with sulfonium ylid reagents<sup>9</sup> showed that the reaction was only successful when the ylide was generated from trimethylsulfonium tetrafluoroborate and *n*-BuLi (Scheme 3). Other sources of dimethylsulfonium ylide or of the corresponding oxide gave complex reaction mixtures.

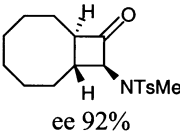
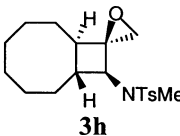
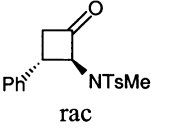
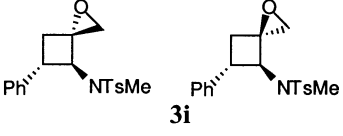
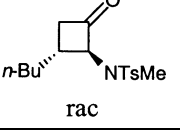
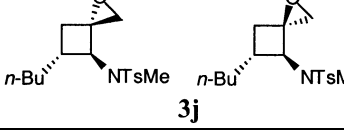
Thus a series of racemic and enantiomerically enriched  $\alpha$ -*N*-methyl *N*-tosyl cyclobutanones **2a–j** (Table 1) previously prepared from the corresponding olefins have been converted into the corresponding oxiranes by treatment with dimethyl sulfonium ylid generated from trimethylsulfonium

Table 1. Preparation of oxiranes 3



Entry	Cyclobutanone 2	Oxirane 3	endo:exo	Yield, % (ee%)
a	 ee 89%	 <b>3a</b>	1:1	69 (89)
b	 ee 92%	 <b>3b</b>	1.5:1	75 (92)
c	 rac	 <b>3c</b>	9:1	86
d	 ee 86%	 <b>3d</b>	6:1	72 (86)
e	 Me NTsMe ee 92%	 Me NTsMe <b>3e</b>	1:4	67 (92)
f	 rac	 <b>3f</b>	2.3:1	82
g	 ee 98%	 <b>3g</b>	–	71 (98)

Table 1 (continued)

Entry	Cyclobutanone 2	Oxirane 3	<i>endo:exo</i>	Yield, % (ee%)
h			–	59 (92)
i			1:9	81
j			1:9	61

tetrafluoroborate and *n*-BuLi (Table 1). *exo*- and *endo*-oxiranes could be easily separated by flash chromatography. Stereochemical assignments by  $^1\text{H}$  NMR were not easy as the protons of the epoxide ring are too far away from the other ring to generate a significant nOe effect. The structure and stereochemistry of crystalline *exo*-**3a**, *exo*-**3b** and *exo*-**3e** were established by X-ray diffraction analysis.<sup>10,11</sup> In these compounds, the hydrogen geminal to the NTsMe group was deshielded with respect to that of the corresponding *endo*-isomers. The stereochemistry of compounds *exo*- and *endo*-**3d** was confirmed by hydrogenation to *exo*- and *endo*-**3b**, respectively (Scheme 4). By analogy, we tentatively assigned relative configurations for all oxiranes **3**.

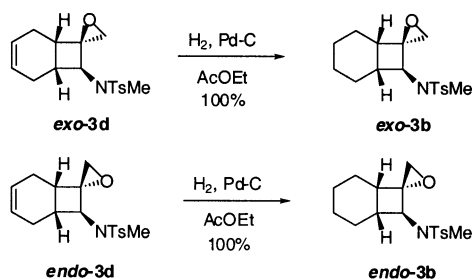
Table 1 shows that oxiranes **3a–j** were formed in good to excellent yields. The facial selectivity can be rationalized by considering the approaches of the ylid reagent to the various cyclobutanones in their preferred conformation. The  $^1\text{H}$  NMR spectra of cyclobutanones **2** showed fairly strong coupling between the proton at the ring junction and the

neighbouring proton geminal to the NTsMe group and was in agreement with previous observations.<sup>12</sup> This indicated that the preferred conformations of the cyclobutanone ring are those shown in Fig. 1.

With *cis*-fused bicyclic ketones **2a,b,d,f** (A, Fig. 1), little selectivity was observed because both  $\alpha$  and  $\beta$  faces are hindered. When the large NTsMe group sits on the  $\alpha$  face as in **2c** (B), selectivity increased in favour of the *endo*-adduct *endo*-**3c**. The presence of a methyl group at the ring junction **2e** (C) further hides the  $\beta$  face and, as a result, the *exo*-adduct became the major isomer. With both the *trans*-fused bicyclic ketones **2g–h** (D) and the mono-substituted cyclobutanones **2i–j** (D), addition occurred preferentially from the  $\alpha$  face.

### 2.3. Rearrangement of oxiranes

A model study of the rearrangement of oxiranes **3** was conducted on each of the two diastereomeric oxiranes *exo*-**3b** and *endo*-**3b** which had previously been prepared from the corresponding enantiomerically enriched bicyclo[4.2.0]octan-7-one **2b** (ee 92%). The *exo*-isomer was selectively



Scheme 4.

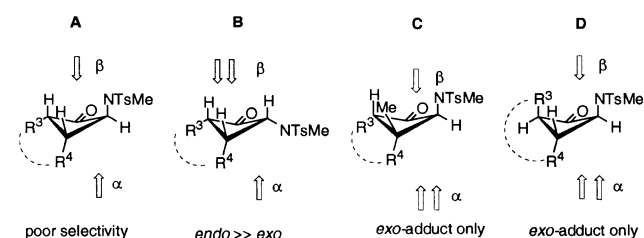
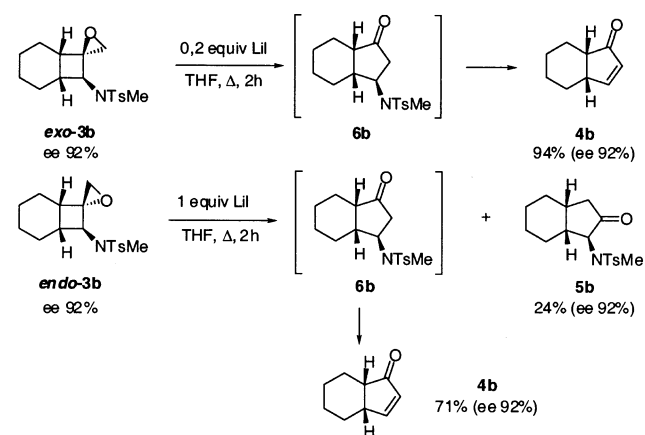
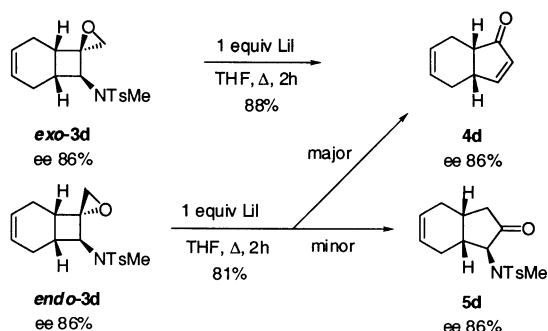


Figure 1.



Scheme 5.



Scheme 6.

converted into (1*S*,6*S*) bicyclo[4.3.0]non-8-en-7-one **4b** (ee 92%) after 2 h in refluxing tetrahydrofuran containing a catalytic amount (0.2 equiv.) of lithium iodide. The formation of **4b** was believed to result from a  $\beta$ -elimination of *N*-methyl-*N*-tosyl amide from bicyclo[4.3.0]nonan-7-one **6b** which was the expected ring-expanded product. This was confirmed by the quantitative formation of **4b** when compound **6b** (previously isolated from the diazomethane ring expansion reaction, see Scheme 2) was refluxed in THF containing lithium iodide.

The *endo*-isomer reacted more slowly: a stoichiometric amount of lithium iodide was needed to bring the rearrangement to completion after 2 hours in refluxing

Table 2. Rearrangement of diastereomeric mixture of 3 in refluxing THF containing a stoichiometric amount of lithium iodide

Entry	Diastereomeric mixture of 3	Enone 4	Yield (%)	ee (%)	Other products
a			95	89	
b			81	92	<b>5b</b> 14%
c			90	rac	
d			41	86	<b>5d</b> 41%
e			95	92	
f			95	rac	
g			90	98	
h			87	92	
i			94	rac	
j			96	rac	

THF. Also the ring expansion was less selective (3:1): the major product was still **4b** but there was a substantial amount of bicyclo[4.3.0]octan-8-one **5b** which is a structural isomer of **6b** (Scheme 5).<sup>13</sup>

The *exo*- and *endo* isomers of the corresponding unsaturated oxiranes *exo*- and *endo*-**3d** showed a similar behaviour (**4d**:**5d**=3:2) (Scheme 6). However all other *exo*- and *endo*-oxiranes gave exclusively the cyclopentenone derivatives in high yields when refluxed in THF containing a stoichiometric amount of lithium iodide. The results shown in Table 2 are those obtained from the mixture of *exo*- and *endo*-oxiranes obtained after flash chromatography of the crude product of the ylid addition to **2**. The rearrangement occurred with retention of configuration at the ring junction except when the five-membered ring was fused to a ring larger than six. A small amount of *trans*-isomer was observed when the cyclopentenone was fused to a seven-membered ring **4f**. Epimerization was completed with the corresponding eight-membered ring derivative **4h**.

Not surprisingly the enantiomeric purities of the original cyclobutanones were retained throughout the sequence.

### 3. Conclusions

In conclusion, the combined results of this and the previous publication<sup>6</sup> provide a unified approach towards the synthesis of enantiomerically pure four and five-membered rings. The previous paper dealt with the enantioselective synthesis of the four-membered ring. The present results showed that these could be readily converted into various types of enantiomerically enriched cyclopentenones. In this [2+2+1] strategy for the cyclopentannulation of olefins, the *NTsMe* group played a crucial role as a control element: as we had anticipated, it acts as an electron-releasing group controlling the regioselectivity of the ring expansion reaction. However, this electronic effect can be counterbalanced by what we believe to be conformational factors as shown in the ring-expansion of cyclobutanones fused to a six-membered ring. At this stage these factors are not well understood. Also, the *NTsMe* group provides the right oxidation state to allow the generation of the conjugate double bond at the end of sequence, thus increasing the synthetic utility of the products.

### 4. Experimental

<sup>1</sup>H NMR spectra were recorded in CDCl<sub>3</sub> on Varian Gemini-200 or 300 spectrometers at 200 or 300 MHz at room temperature. <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub> at 50 or 75 MHz at room temperature. Chemical shifts are given in ppm relative to (CH<sub>3</sub>)<sub>4</sub>Si (0 ppm, <sup>1</sup>H) or CDCl<sub>3</sub> (77.0 ppm, <sup>13</sup>C). Mass spectra were obtained on a Finnigan MAT-TSQ 700 spectrometer. IR spectra were recorded on a BIO-RAD TFS 135 FT-IR spectrometer. All absorption values are expressed in wavenumbers (cm<sup>-1</sup>). [α]<sub>D</sub> values were obtained on a Perkin–Elmer 241 MC polarimeter. Melting points are uncorrected. Enantiomeric excesses were measured on hplc with a Millipore Waters 600 Controller, UV Millipore Waters 486 as detector and

fitted with Diacel Chiralpack-AD, -AS, -OB, -OD analytical column. TLC were run on silica gel 60F<sub>254</sub>. Column chromatographies were performed with gel 40 (230–400 μm, Merck). All solvents were distilled before use. All reagents were of reagent grade.

#### 4.1. General procedure for the preparation of epoxides

A solution of *n*-BuLi (2.5 M in hexanes, 1.2 equiv.) was added to a 0.05 M solution of trimethylsulfonium tetrafluoroborate (1 equiv.) in THF at –15°C. After 15 min, the mixture was cooled to –78°C and treated with a solution of cyclobutanone (1 equiv.) in THF. After 15 h at room temperature, the reaction mixture was diluted with ether and washed twice with water. The organic layer was dried over MgSO<sub>4</sub>, filtered and concentrated. The product was purified by flash chromatography on silica gel (eluent AcOEt/cyclohexane 2:8).

#### 4.2. Preparation of epoxides *exo*- and *endo*-**3a**

From **2a** (250 mg, 0.85 mmol), trimethylsulfonium tetrafluoroborate (168 mg, 1.02 mmol), *n*-BuLi (2.0 M in hexanes, 640 μL, 1.28 mmol). Yield: 85 mg (33%) of *exo*-**3a** and 95 mg (36%) of *endo*-**3a** (89% ee).

**4.2.1. (1*R*,5*S*,6*R*,7*S*)-7-(Methyltosylamino)spiro[bicyclo[3.2.0]heptane-6,2'-oxirane] (*exo*-**3a**).** Mp 88°C; [α]<sub>D</sub><sup>20</sup>=+38.8 (*c*=0.6, CHCl<sub>3</sub>); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.65 (d, *J*=8.4 Hz, 2H), 7.29 (d, *J*=8.4 Hz, 2H), 4.29 (dd, *J*=4.2, 1.6 Hz, 1H), 3.06 (s, 3H), 2.87 (t, *J*=8.0 Hz, 1H), 2.71 (td, *J*=8.0, 4.4 Hz, 1H), 2.53 (d, *J*=5.2 Hz, 1H), 2.42 (s, 3H), 2.30 (d, *J*=5.2 Hz), 1.90–1.20 (m, 6H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) δ 143.1, 136.4, 129.4, 127.1, 62.9, 62.4, 48.5, 45.1, 38.7, 31.5, 30.8, 29.4, 25.6, 21.5; IR (neat) ν 2943, 2851, 1338, 1157; M (IE): 238 (100%), 155 (56%), 91 (72%); elemental analysis calcd (%) for C<sub>16</sub>H<sub>21</sub>O<sub>3</sub>NS: C 62.52, H 6.88, N 4.56, S 10.43; found: C 62.19, H 6.83, N 4.48, S 10.13; HPLC (AD column; eluent EtOH; flow: ν=0.35 mL/min; λ=254 nm): 15.4 min (1*S*,5*R*,6*R*,7*R*) and 21.6 min (1*R*,5*S*,6*S*,7*S*).

**4.2.2. (1*R*,5*S*,6*S*,7*S*)-7-(Methyltosylamino)spiro[bicyclo[3.2.0]heptane-6,2'-oxirane] (*endo*-**3a**).** [α]<sub>D</sub><sup>20</sup>=+29.2 (*c*=1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.66 (d, *J*=8.4 Hz, 2H), 7.29 (d, *J*=8.4 Hz, 2H), 4.31 (dd, *J*=4.7, 1.6 Hz, 1H), 2.88 (s, 3H), 2.88–2.80 (m, 1H), 2.65–2.55 (m, 1H), 2.59 (d, *J*=5.3 Hz, 1H), 2.43 (s, 3H), 2.29 (d, *J*=5.3 Hz, 1H), 1.90–1.20 (m, 6H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) δ 143.3, 136.9, 129.5, 127.2, 63.7, 61.0, 51.1, 41.6, 37.8, 31.9, 30.7, 26.6, 25.4, 21.5; IR (neat) ν 2950, 1340, 1159; M (IE): 308 (24%), 238 (100%), 155 (36%); HRMS calcd (%) for C<sub>16</sub>H<sub>21</sub>O<sub>3</sub>NS: 307.1242; found: 307.1249; HPLC (AD column; eluent EtOH/hexane 10:90; flow: ν=0.50 mL/min; λ=220 nm): 32.1 min (1*S*,5*R*,6*S*,7*R*) and 36.9 min (1*R*,5*S*,6*R*,7*S*).

#### 4.3. Preparation of epoxides *exo*- and *endo*-**3b**

From **2b** (3.3 g, 10.8 mmol), trimethylsulfonium tetrafluoroborate (2 g, 12.3 mmol), *n*-BuLi (6.5 mL, 16.2 mmol). Yield: 0.95 g (47%) of *exo*-**3b** and 0.58 g (29%) of *endo*-**3b** (92% ee).

**4.3.1. (1R,6S,7R,8S)-8-(Methyltosylamino)spiro[bicyclo[4.2.0]octane-7,2'-oxirane] (exo-3b).**  $[\alpha]_D^{20} = +10.6$  ( $c=0.85$ ,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.61 (d,  $J=8.3$  Hz, 2H), 7.22 (d,  $J=8.3$  Hz, 2H), 4.61 (dd,  $J=8.3$ , 1.4 Hz, 1H), 2.91 (s, 3H), 2.58–2.70 (m, 1H), 2.55 (d,  $J=4.6$  Hz, 1H), 2.36 (d,  $J=4.6$  Hz, 1H), 2.42 (s, 3H), 2.25 (m, 1H), 1.70–0.70 (m, 8H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.2, 136.7, 129.5, 126.9, 66.5, 57.5, 46.8, 35.6, 33.4, 31.4, 24.9, 24.1, 22.3, 21.5, 21.4; IR (neat)  $\nu$  2927, 2854, 1340, 1158; M (IE): 322 (2%), 238 (16%), 185 (24%), 155 (100%); 91 (72%); elemental analysis calcd (%) for  $\text{C}_{17}\text{H}_{23}\text{O}_3\text{NS}$ : C 63.52, H 7.21, N 4.35, S 9.97; found: C 63.43, H 7.34, N 4.13, S 10.04; HPLC (AD column; eluent EtOH; flow:  $\nu=0.35$  mL/min;  $\lambda=254$  nm): 14.1 min (1S,6R,7S,8R) and 22.0 min (1R,6S,7R,8S)

**4.3.2. (1R,6S,7S,8S)-8-(Methyltosylamino)spiro[bicyclo[4.2.0]octane-7,2'-oxirane] (endo-3b).** Mp 104°C;  $[\alpha]_D^{20} = -0.95$  ( $c=0.85$ ,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.70 (d,  $J=8.2$  Hz, 2H), 7.28 (d,  $J=8.2$  Hz, 2H), 4.83 (d,  $J=9.1$  Hz, 1H), 2.79 (s, 3H), 2.72 (d,  $J=9.0$  Hz, 1H), 2.49 (d,  $J=9.0$  Hz, 1H), 2.42 (s, 3H), 2.38–2.26 (m, 2H), 1.90–0.90 (m, 8H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.3, 136.4, 129.4, 127.2, 63.1, 57.9, 50.1, 32.9, 30.4, 29.6, 23.2, 22.4, 21.4, 21.3; IR (neat)  $\nu$  2929, 2852, 1340, 1161; M (IE): 321 (4%), 238 (72%), 166 (36%), 155 (32%), 91 (60%), 42 (100%); elemental analysis calcd (%) for  $\text{C}_{17}\text{H}_{23}\text{O}_3\text{NS}$ : C 63.52, H 7.21, N 4.35, S 9.97; found: C 63.46, H 7.25, N 4.29, S 10.35; HPLC (AD column; eluent EtOH/hexane 10:90; flow:  $\nu=1.00$  mL/min;  $\lambda=254$  nm): 12.7 min (1R,6S,7R,8S) and 14.4 min (1S,6R,7S,8R).

#### 4.4. Preparation of epoxides *exo*- and *endo*-3c

From **2c** (500 mg, 1.6 mmol), trimethylsulfonium tetrafluoroborate (320 mg, 1.9 mmol), *n*-BuLi (975  $\mu\text{L}$ , 2.4 mmol). Yield: 40 mg (8%) of *exo*-**3c** and 410 mg (78%) of *endo*-**3c**.

**4.4.1. (1R\*,6S\*,7R\*,8R\*)-8-(Methyltosylamino)spiro[bicyclo[4.2.0]octane-7,2'-oxirane] (exo-3c).**  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.71 (d,  $J=8.4$  Hz, 2H), 7.28 (d,  $J=8.4$  Hz, 2H), 4.83 (d,  $J=8.5$  Hz, 1H), 2.79 (s, 3H), 2.72 (d,  $J=5.1$  Hz, 1H), 2.50 (d,  $J=5.1$  Hz, 1H), 2.50–2.20 (m, 2H), 2.42 (s, 3H), 1.80–0.95 (m, 8H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.3, 136.6, 129.5, 127.2, 63.2, 58.0, 50.1, 33.0, 30.4, 29.4, 23.2, 23.1, 22.4, 21.4, 21.3; IR (neat)  $\nu$  2933, 2851, 1341, 1160; M (IE): 322 (2%), 238 (100%), 166 (54%), 155 (40%), 91 (46%), 42 (48%).

**4.4.2. (1R\*,6S\*,7S\*,8R\*)-8-(Methyltosylamino)spiro[bicyclo[4.2.0]octane-7,2'-oxirane] (endo-3c).** Mp 148°C;  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67 (d,  $J=8.2$  Hz, 2H), 7.31 (d,  $J=8.2$  Hz, 2H), 4.25 (d,  $J=9.1$  Hz, 1H), 2.79 (s, 3H), 2.72–2.50 (m, 2H), 2.44 (s, 3H), 2.42 (d,  $J=9.1$  Hz, 1H), 1.85–0.95 (m, 4H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.3, 134.2, 129.5, 127.6, 68.5, 55.6, 44.5, 39.9, 33.3, 32.8, 23.8, 22.3, 22.1, 21.4, 19.3; IR (neat)  $\nu$  2927, 2842, 1340, 1158; M (IE): 322 (2%), 238 (100%), 166 (46%), 155 (36%), 91 (50%), 42 (44%); elemental analysis calcd (%) for  $\text{C}_{17}\text{H}_{23}\text{O}_3\text{NS}$ : C 63.52, H 7.21, N 4.35, S 9.97; found: C 63.49, H 7.37, N 4.23, S 10.26.

#### 4.5. Preparation of epoxides *exo*- and *endo*-3d

From **2d** (3.1 g, 10.2 mmol), trimethylsulfonium tetrafluoroborate (2 g, 12.3 mmol), *n*-BuLi (6.1 mL, 16.2 mmol). Yield: 338 mg (10%) of *exo*-**3d** and 2.02 g (62%) of *endo*-**3d** (86% ee).

**4.5.1. (1R,6S,7R,8S)-8-(Methyltosylamino)spiro[bicyclo[4.2.0]oct-3-ene-7,2'-oxirane] (exo-3d).**  $[\alpha]_D^{20} = -5.5$  ( $c=0.11$ ,  $\text{CH}_2\text{Cl}_2$ );  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.64 (d,  $J=8.4$  Hz, 2H), 7.28 (d,  $J=8.4$  Hz, 2H), 5.95–5.85 (m, 2H), 4.38 (dd,  $J=6.1$ , 1.1 Hz, 1H), 3.06 (s, 3H), 2.90–2.65 (m, 2H), 2.53 (d,  $J=4.8$  Hz, 1H), 2.42 (s, 3H), 2.22 (d,  $J=4.8$  Hz, 1H), 2.10–1.80 (m, 4H);  $^{13}\text{C NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  143.2, 136.9, 129.5, 127.9, 127.6, 127.0, 64.3, 61.1, 48.2, 37.1, 32.9, 31.6, 25.0, 23.4, 21.5; IR (neat)  $\nu$  3031, 2923, 2830, 1598, 1338, 1160; M (IE): 320 (1%), 212 (96%), 164 (28%), 91 (32%), 43 (100%); elemental analysis calcd (%) for  $\text{C}_{17}\text{H}_{21}\text{O}_3\text{NS}$ : C 63.93, H 6.63, N 4.39, S 10.04; found: C 63.80, H 6.58, N 4.38, S 10.24.

**4.5.2. (1R,6S,7S,8S)-8-(Methyltosylamino)spiro[bicyclo[4.2.0]oct-3-ene-7,2'-oxirane] (endo-3d).** Mp 123°C;  $[\alpha]_D^{20} = +18.1$  ( $c=0.70$ ,  $\text{CH}_2\text{Cl}_2$ );  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.66 (d,  $J=8.2$  Hz, 2H), 7.28 (d,  $J=8.2$  Hz, 2H), 6.00–5.90 (m, 1H), 5.88–5.80 (m, 1H), 4.60 (d,  $J=6.5$  Hz, 1H), 2.86 (s, 3H), 2.72–2.57 (m, 2H), 2.64 (d,  $J=5.1$  Hz, 1H), 2.42 (s, 3H), 2.32 (d,  $J=5.1$  Hz, 1H), 2.10–1.70 (m, 4H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.3, 136.3, 129.4, 127.7, 127.1, 125.5, 62.3, 61.7, 50.8, 32.5, 30.8, 29.8, 23.9, 21.4, 20.4; IR (neat)  $\nu$  3030, 2963, 2836, 1598, 1339, 1159; M (IE): 319 (4%), 238 (36%), 164 (100%), 155 (52%), 91 (80%); HRMS calcd (%) for  $\text{C}_{17}\text{H}_{21}\text{O}_3\text{NS}$ : 319.1242; found: 319.1240.

#### 4.6. Preparation of epoxides *exo*- and *endo*-3e

From **2e** (250 mg, 0.85 mmol), trimethylsulfonium tetrafluoroborate (168 mg, 1.02 mmol), *n*-BuLi (610  $\mu\text{L}$ , 1.5 mmol). Yield: 176 mg (54%) of *exo*-**3e** and 44 mg (13%) of *endo*-**3e** (92% ee).

**4.6.1. (1R,5S,6R,7S)-1-Methyl-7-(methyltosylamino)spiro[bicyclo[3.2.0]heptane-6,2'-oxirane] (exo-3e).**  $[\alpha]_D^{20} = +37.9$  ( $c=0.9$ ,  $\text{CH}_2\text{Cl}_2$ );  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.63 (d,  $J=7.8$  Hz, 2H), 7.27 (d,  $J=7.8$  Hz, 2H), 4.31 (d,  $J=1.5$  Hz, 1H), 3.13 (s, 3H), 2.48 (d,  $J=5.0$  Hz, 1H), 2.50–2.40 (m, 1H), 2.42 (s, 3H), 2.14 (d,  $J=5.0$  Hz, 1H), 1.95–1.20 (m, 6H), 1.21 (s, 3H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.1, 137.0, 129.4, 126.8, 64.6, 61.8, 51.8, 48.3, 48.1, 40.4, 34.4, 29.5, 26.2, 21.4, 20.8; IR (neat)  $\nu$  2957, 2869, 1339, 1260, 1157; M (IE): 321 (4%), 239 (44%), 155 (28%), 91 (56%), 42 (100%); elemental analysis calcd (%) for  $\text{C}_{17}\text{H}_{23}\text{O}_3\text{NS}$ : C 63.52, H 7.21, N 4.36, S 9.97; found: C 63.38, H 6.99, N 4.07, S 9.97; HPLC (AD column; eluent EtOH; flow:  $\nu=0.35$  mL/min;  $\lambda=254$  nm): 13.2 min (1S,5R,6R,7R) and 15.8 min (1R,5S,6S,7S).

**4.6.2. (1R,5S,6S,7S)-1-Methyl-7-(methyltosylamino)spiro[bicyclo[3.2.0]heptane-6,2'-oxirane] (endo-3e).**  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.65 (d,  $J=8.1$  Hz, 2H), 7.29 (d,  $J=8.1$  Hz, 2H), 4.44 (d,  $J=1.6$  Hz, 1H), 2.93 (s, 3H), 2.66 (d,  $J=5.1$  Hz, 1H), 2.50–2.40 (m, 1H), 2.42 (s, 3H),

2.31 (d,  $J=5.1$  Hz, 1H), 2.00–1.10 (m, 6H), 1.18 (s, 3H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.2, 136.3, 129.5, 127.1, 66.0, 60.3, 52.1, 48.6, 48.0, 41.7, 33.6, 27.2, 26.4, 21.5, 20.4; IR (neat)  $\nu$  2957, 2869, 1339, 1260, 1157; M (IE): 321 (12%), 185 (60%), 155 (40%), 91 (100%); HRMS calcd (%) for  $\text{C}_{17}\text{H}_{23}\text{O}_3\text{NS}$ : 321.1402; found: 321.1399.

#### 4.7. Preparation of epoxides *exo*- and *endo*-3f

From **2f** (400 mg, 1.2 mmol), trimethylsulfonium tetrafluoroborate (245 mg, 1.5 mmol), *n*-BuLi (2.3 M in hexanes, 830  $\mu\text{L}$ , 1.9 mmol). Yield: 84 mg (20%) of *exo*-**3f** and 260 mg (62%) of *endo*-**3f**.

**4.7.1. (1R\*,7S\*,8R\*,9S\*)-9-(Methyltosylamino)spiro[bicyclo[5.2.0]nonane-8,2'-oxirane] (*exo*-3f).**  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.65 (d,  $J=8.2$  Hz, 2H), 7.28 (d,  $J=8.2$  Hz, 2H), 4.45 (dd,  $J=6.8, 1.2$  Hz, 1H), 3.03 (s, 3H), 2.80–2.50 (m, 2H), 2.55 (d,  $J=4.9$  Hz, 1H), 2.42 (s, 3H), 2.29 (d,  $J=4.9$  Hz, 1H), 1.90–1.00 (m, 10H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.2, 137.0, 129.5, 127.0, 63.6, 60.5, 47.8, 44.0, 39.8, 31.9, 31.5, 31.4, 29.1, 28.7, 28.3, 21.5; IR (neat)  $\nu$  2990, 2918, 2847, 1376, 1160; M (IE): 335 (4%), 238 (100%), 180 (40%), 155 (28%), 91 (32%), 42 (26%); HRMS (CI) calcd (%) for  $\text{C}_{18}\text{H}_{26}\text{O}_3\text{NS}$ : 336.1633; found: 336.1627.

**4.7.2. (1R\*,7S\*,8S\*,9S\*)-9-(Methyltosylamino)spiro[bicyclo[5.2.0]nonane-8,2'-oxirane] (*endo*-3f).**  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67 (d,  $J=8.2$  Hz, 2H), 7.28 (d,  $J=8.2$  Hz, 2H), 4.56 (d,  $J=6.3$  Hz, 1H), 2.83 (s, 3H), 2.60 (d,  $J=5.0$  Hz, 1H), 2.50–2.35 (m, 2H), 2.42 (s, 3H), 2.33 (d,  $J=5.0$  Hz, 1H), 2.00–1.00 (m, 10H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.2, 136.2, 129.4, 127.2, 63.5, 61.0, 50.0, 41.5, 36.9, 32.0, 30.7, 29.8, 28.1, 25.9, 21.4; IR (neat)  $\nu$  2918, 2847, 1337, 1151; M (IE): 335 (4%), 238 (100%), 180 (26%), 155 (20%), 150 (8%), 91 (30%), 42 (54%).

**4.7.3. Preparation of epoxide (1R,7R,8R,9S)-9-(methyltosylamino)spiro[bicyclo[5.2.0]nonane-8,2'-oxirane] (*exo*-3g).** From **2g** (348 mg, 1.1 mmol), trimethylsulfonium tetrafluoroborate (213 mg, 1.3 mmol), *n*-BuLi (2.1 M in hexanes, 785  $\mu\text{L}$ , 1.6 mmol). Yield: 268 mg (74%);  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67 (d,  $J=8.3$  Hz, 2H), 7.29 (d,  $J=8.3$  Hz, 2H), 4.34 (d,  $J=9.4$  Hz, 1H), 2.93 (s, 3H), 2.51 (s, 2H), 2.42 (s, 3H), 2.45–2.30 (m, 1H), 2.30–2.15 (m, 1H), 1.70–1.10 (m, 10H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.1, 136.8, 129.5, 127.0, 66.9, 57.9, 45.9, 43.8, 40.6, 31.5, 31.3, 29.0, 28.0, 25.3, 23.5, 21.4; IR (neat)  $\nu$  2918, 2856, 1342, 1156; M (IE): 335 (4%), 238 (100%), 180 (26%), 155 (24%), 150 (10%), 91 (42%), 42 (52%); HRMS calcd (%) for  $\text{C}_{18}\text{H}_{25}\text{O}_3\text{NS}$ : 335.1555; found: 336.1555.

**4.7.4. Preparation of epoxide (1R,8R,9R,10S)-10-(methyltosylamino)spiro[bicyclo[6.2.0]decane-9,2'-oxirane] (*exo*-3h).** From **2h** (431 mg, 1.3 mmol), trimethylsulfonium tetrafluoroborate (253 mg, 1.5 mmol), *n*-BuLi (2.0 M in hexanes, 965  $\mu\text{L}$ , 1.9 mmol). Yield: 264 mg (59%); mp 85°C;  $[\alpha]_D^{20} = -30.7$  ( $c=0.8$ ,  $\text{CH}_2\text{Cl}_2$ );  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67 (d,  $J=8.2$  Hz, 2H), 7.28 (d,  $J=8.2$  Hz, 2H), 4.31 (d,  $J=9.3$  Hz, 1H), 2.96 (s, 3H), 2.45 (d,  $J=4.5$  Hz, 1H), 2.42 (s, 3H), 2.36 (d,  $J=4.5$  Hz, 1H),

2.30–2.15 (m, 1H), 1.90–1.65 (m, 4H), 1.60–1.45 (m, 1H), 1.40–1.10 (m, 7H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  143.1, 137.0, 129.5, 127.0, 66.6, 59.2, 46.1, 44.6, 40.9, 34.4, 31.7, 28.2, 27.4, 27.0, 26.8, 26.0, 21.4; IR (neat)  $\nu$  2918, 2847, 1337, 1147; M (IE): 349 (12%), 306 (4%), 238 (100%), 194 (32%), 155 (24%), 91 (36%), 42 (72%); elemental analysis calcd (%) for  $\text{C}_{19}\text{H}_{27}\text{O}_3\text{NS}$ : C 65.30, H 7.74, N 4.01, S 9.17; found: C 65.14, H 7.85, N 3.90, S 9.80.

#### 4.8. Preparation of epoxides *exo*- and *endo*-3i

From **2i** (146 mg, 0.4 mmol), trimethylsulfonium tetrafluoroborate (73 mg, 0.4 mmol), *n*-BuLi (2.0 M in hexanes, 266  $\mu\text{L}$ , 0.5 mmol). Yield: 111 mg (73%) of *exo*-**3i** and 12 mg (8%) of *endo*-**3i**.

**4.8.1. (1R\*,2S\*,3R\*)-2-(Methyltosylamino)-3-phenylspiro[cyclobutane-1,2'-oxirane] (*exo*-3i).**  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.40 (d,  $J=8.1$  Hz, 2H), 7.30–7.00 (m, 5H), 7.00 (d,  $J=8.1$  Hz, 2H), 4.96 (d,  $J=9.5$  Hz, 1H), 3.89 (q,  $J=9.5$  Hz, 1H), 3.10 (s, 3H), 2.70 (d,  $J=4.3$  Hz, 1H), 2.55 (d,  $J=4.3$  Hz, 1H), 2.48 (d,  $J=9.5$  Hz, 1H), 2.46 (d,  $J=9.5$  Hz, 1H), 2.35 (s, 3H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  142.9, 141.3, 136.5, 129.4, 128.6, 126.9, 126.6, 126.5, 63.3, 62.2, 47.8, 41.5, 33.0, 31.6, 21.4; IR (neat)  $\nu$  2923, 2855, 1340, 1166; M (IE): 343 (2%), 287 (76%), 239 (36%), 188 (80%), 155 (32%), 106 (98%), 91 (100%), 42 (34%); HRMS calcd (%) for  $\text{C}_{19}\text{H}_{21}\text{O}_3\text{NS}$ : 343.1242; found: 343.1254.

**4.8.2. (1S\*,2S\*,3R\*)-2-(Methyltosylamino)-3-phenylspiro[cyclobutane-1,2'-oxirane] (*endo*-3i).** Mp 134°C;  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.44 (d,  $J=8.2$  Hz, 2H), 7.30 (m, 5H), 7.05 (d,  $J=8.2$  Hz, 2H), 5.00 (d,  $J=9.3$  Hz, 1H), 3.40 (q,  $J=9.3$  Hz, 1H), 2.91 (s, 3H), 2.81 (d,  $J=5.2$  Hz, 1H), 2.60 (d,  $J=5.2$  Hz, 1H), 2.52–2.40 (m, 2H), 2.35 (s, 3H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.0, 141.0, 136.5, 129.3, 128.6, 127.0, 126.7, 126.7, 64.7, 59.5, 49.6, 36.3, 31.8, 30.9, 21.4; IR (neat)  $\nu$  2926, 2855, 1339, 1166; M (IE): 343 (2%), 287 (92%), 239 (24%), 188 (96%), 155 (32%), 91 (100%), 42 (48%).

#### 4.9. Preparation of epoxides *exo*- and *endo*-3j

From **2j** (500 mg, 1.6 mmol), trimethylsulfonium tetrafluoroborate (300 mg, 1.9 mmol), *n*-BuLi (1.9 M in hexanes, 935  $\mu\text{L}$ , 1.8 mmol). Yield: 320 mg (61%) of *exo*-**3j** and 25 mg (5%) of *endo*-**3j**.

**4.9.1. (1R\*,2S\*,3R\*)-3-Butyl-2-(methyltosylamino)spiro[cyclobutane-1,2'-oxirane] (*exo*-3j).**  $^1\text{H}$  NMR (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67 (d,  $J=8.4$  Hz, 2H), 7.29 (d,  $J=8.4$  Hz, 2H), 4.40 (d,  $J=8.0$  Hz, 1H), 3.03 (s, 3H), 2.80–2.40 (m, 1H), 2.54 (d,  $J=4.6$  Hz, 1H), 2.42 (s, 3H), 2.32 (d,  $J=4.6$  Hz, 1H), 2.15 (ddd,  $J=13.3, 9.8, 1.3$  Hz, 1H), 1.92 (dd,  $J=13.3, 9.8$  Hz, 1H), 1.45–1.00 (m, 6H), 0.92 (t,  $J=6.6$  Hz, 3H);  $^{13}\text{C}$  NMR (50 MHz,  $\text{CDCl}_3$ )  $\delta$  143.2, 137.2, 129.5, 127.0, 62.6, 60.8, 48.3, 36.8, 34.3, 31.7, 31.6, 29.3, 22.5, 21.4, 13.8; IR (neat)  $\nu$  2959, 2930, 1348, 1251, 1157; M (IE): 323 (6%), 267 (100%), 238 (42%), 168 (24%), 155 (48%), 91 (45%); elemental analysis calcd (%) for  $\text{C}_{17}\text{H}_{25}\text{O}_3\text{NS}$ : C 63.13, H 7.79, N 4.33, S 9.91; found: C 63.11, H 7.78, N 4.13, S 9.66.

**4.9.2. (1*S*\*,2*S*\*,3*R*\*)-3-Butyl-2-(methyltosylamino)spiro[cyclobutane-1,2'-oxirane] (*endo*-**3j**).** <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>) δ 7.69 (d, *J*=8.1 Hz, 2H), 7.28 (d, *J*=8.1 Hz, 2H), 4.49 (d, *J*=7.2 Hz, 1H), 2.81 (s, 3H), 2.65 (d, *J*=4.9 Hz, 1H), 2.42 (s, 3H), 2.33 (d, *J*=4.9 Hz, 1H), 2.25–2.10 (m, 2H), 1.90–1.70 (m, 1H), 1.50–1.30 (m, 4H), 1.40 (q, *J*=6.7 Hz, 2H), 0.84 (t, *J*=6.7 Hz, 3H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) δ 143.2, 136.6, 129.4, 127.2, 63.2, 59.5, 49.8, 34.5, 32.1, 30.8, 30.4, 29.3, 22.5, 21.4, 13.8.

#### 4.10. Reaction of **2b** with diazomethane

A 1 M solution of diazomethane in ether (10 mL) was added to a solution of **2b** (350 mg, 1.2 mmol) in a 1:1 mixture of ether and ethyle acetate at –10°C. After one day, drops of acetic acid were added and the mixture was concentrated in vacuo and purified by flash chromatography on silica gel (eluent AcOEt/cyclohexane 2:8). Yield 75 mg (19%) of **6** and 20 mg (5%) of **5** and 245 mg (70%) of starting material **2b**.

**4.10.1. (1*R*\*,6*S*\*,9*S*\*)-9-(Methyltosylamino)-bicyclo[4.3.0]nonan-7-one (**6b**).** <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.67 (d, *J*=7.9 Hz, 2H), 7.29 (d, *J*=7.9 Hz, 2H), 4.59 (q, *J*=9.0 Hz, 1H), 2.71 (s, 3H), 2.41 (s, 3H), 2.32 (m, 1H), 2.22 (m, 1H), 2.14 (dd, *J*=19.0, 8.0 Hz, 1H), 1.95 (ddd, *J*=19.0, 9.0, 1.0 Hz, 1H), 1.68–1.58 (m, 3H), 1.50–1.37 (m, 3H), 1.33–1.15 (m, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 215.3, 143.5, 136.1, 129.8, 127.0, 53.8, 47.8, 38.0, 37.4, 28.0, 23.8, 23.2, 22.9, 21.4, 20.7; IR (neat) ν 2928, 2856, 1742, 1339, 1159; M (IC+): 322 (2%), 186 (100%); HRMS calcd (%) for C<sub>17</sub>H<sub>23</sub>O<sub>3</sub>NS: 321.1397; found: 321.1392.

**4.10.2. (1*S*\*,6*R*\*,7*S*\*)-7-(Methyltosylamino)bicyclo[4.3.0]nonan-8-one (**5b**).** Mp 104°C; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.79 (d, *J*=8.0 Hz, 2H), 7.31 (d, *J*=8.0 Hz, 2H), 4.66 (d, *J*=12.4 Hz, 1H), 2.59 (s, 3H), 2.43 (s, 3H), 2.29 (m, 1H), 2.25 (m, 2H), 2.02 (m, 1H), 1.92 (m, 1H), 1.76 (m, 1H), 1.72 (m, 1H), 1.59 (m, 2H), 1.57 (m, 1H), 1.28 (m, 1H), 1.14 (m, 1H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 213.5, 143.2, 136.8, 129.4, 127.6, 63.7, 43.9, 36.7, 31.4, 30.0, 29.2, 25.0, 21.4, 19.6; IR (neat) ν 2927, 2854, 1749, 1338, 1156; M (IE): 321 (28%), 165 (100%), 136 (100%), 91 (56%); elemental analysis calcd (%) for C<sub>17</sub>H<sub>23</sub>O<sub>3</sub>NS: C 63.52, H 7.21, N 4.36, S 9.97; found: C 63.27, H 7.16, N 4.22, S 10.14.

#### 4.11. General procedure for the rearrangement of epoxides

Lithium iodide (1 equiv.) was added to a 0.05 M solution of epoxide (1 equiv.) in THF at room temperature. The mixture was heated for two hours at 65°C. The mixture was cooled to room temperature and concentrated in vacuo. The product was purified by flash chromatography on silica gel (eluent AcOEt/cyclohexane 2:8). When the separation of oxiranes was not required, the crude reaction mixture was purified by flash chromatography to afford a mixture of epoxides. The rearrangement was done on these mixtures.

#### 4.12. Rearrangement of **3a**

From *exo*-**3a** and *endo*-**3a** (400 mg, 1.2 mmol), lithium iodide (170 mg, 1.2 mmol). Yield: 130 mg (86%) of **4a**

(90% ee). (1*S*,5*S*)-Bicyclo[3.3.0]oct-3-en-2-one (**4a**): RN 1388665-76-6; [α]<sub>D</sub><sup>20</sup>=+291.5 (*c*=1.1, CH<sub>2</sub>Cl<sub>2</sub>); HPLC (OB column; eluent *i*PrOH/hexane 2:98; flow: ν=1.00 mL/min; λ=232 nm): 10.6 min (1*R*,5*R*) and 12.2 min (1*S*,5*S*).

#### 4.13. Rearrangement of **2b**

From *exo*-**2b** (200 mg, 0.62 mmol), lithium iodide (83 mg, 0.62 mmol). Yield: 80 mg (95%) of **4b** (92% ee). From *endo*-**2b** (200 mg, 0.62 mmol), lithium iodide (8 mg, 0.62 mmol). Yield: 60 mg (71%) of **4b** and 48 mg (24%) of **5b** (92% ee).

(1*S*,6*S*)-Bicyclo[4.3.0]non-8-en-7-one (**4b**): RN 81255-91-6; [α]<sub>D</sub><sup>20</sup>=+35.3 (*c*=0.1, CH<sub>2</sub>Cl<sub>2</sub>); HPLC (OB column; eluent *i*PrOH/hexane 2:98; flow: ν=1.00 mL/min; λ=232 nm): 8.0 min (1*S*,6*S*) and 17.0 min (1*R*,6*R*).

**4.13.1. (1*S*,6*R*,7*S*)-7-(Methyltosylamino)bicyclo[4.3.0]nonan-8-one (**5b**).** [α]<sub>D</sub><sup>20</sup>=+93.4 (*c*=1.1, CHCl<sub>3</sub>); HPLC (AD column; eluent EtOH; flow: ν=0.35 mL/min; λ=254 nm): 17.9 min (1*R*,6*S*,7*R*) and 30.9 min (1*S*,6*R*,7*S*).

#### 4.14. Rearrangement of **3c**

From *exo*-**3c**-rac (100 mg, 0.3 mmol), lithium iodide (27 mg, 0.3 mmol) heated overnight. Yield: 36 mg (90%) of **4a**.

#### 4.15. Rearrangement of **3d**

From *exo*-**3d** (100 mg, 0.31 mmol), lithium iodide (42 mg, 0.31 mmol). Yield: 37 mg (88%) of **4d** (86% ee). From *endo*-**3d** (1 g, 3.13 mmol), lithium iodide (419 mg, 3.13 mmol). Yield: 205 mg (32%) of **4d** and 320 mg (49%) of **5d** (86% ee).

**4.15.1. (1*S*,6*S*)-Bicyclo[4.3.0]non-3,8-dien-7-one (**4d**).** RN 94344-47-5; [α]<sub>D</sub><sup>20</sup>=+99.5 (*c*=0.6, CH<sub>2</sub>Cl<sub>2</sub>); HPLC (OB column; eluent *i*PrOH/hexane 2:98; flow: ν=1.00 mL/min; λ=220 nm): 11.9 min (1*S*,6*S*) and 19.5 min (1*R*,6*R*).

**4.15.2. (1*S*,6*R*,7*S*)-7-(Methyltosylamino)bicyclo[4.3.0]non-3-en-8-one (**5d**).** <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>) δ 7.75 (d, *J*=8.4 Hz, 2H), 7.30 (d, *J*=8.4 Hz, 2H), 5.80–5.70 (m, 2H), 4.33 (d, *J*=11.9 Hz, 1H), 2.65 (s, 3H), 2.43 (s, 3H), 2.50–2.15 (m, 6H), 1.80–1.55 (m, 2H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>) δ 212.7, 143.0, 136.7, 129.2, 127.2, 124.3, 123.9, 65.6, 44.6, 35.0, 30.1, 27.2, 26.8, 23.4, 21.2; IR (neat) ν 2922, 2840, 1749, 1338, 1153; M (IE): 319 (2%), 185 (28%), 155 (16%), 91 (26%), 43 (100%); HRMS calcd (%) for C<sub>17</sub>H<sub>21</sub>O<sub>3</sub>NS: 319.1242; found: 319.1247; [α]<sub>D</sub><sup>20</sup>=+101.7 (*c*=0.7, CH<sub>2</sub>Cl<sub>2</sub>); HPLC (AD column; eluent EtOH; flow: ν=0.35 mL/min; λ=254 nm): 16.4 min (1*R*,6*S*,7*R*) and 21.0 min (1*S*,6*R*,7*S*).

#### 4.16. Rearrangement of **3e**

From *exo*-**3e** and *endo*-**3e** (400 mg, 1.2 mmol), lithium iodide (170 mg, 1.2 mmol). Yield: 160 mg (95%) of **4e** (92% ee).



**4.16.1. (1*S*,5*S*)-5-Methyl-bicyclo[3.3.0]oct-3-en-2-one (4e).** RN 39163-28-5;  $[\alpha]_D^{20} = +14.8$  ( $c = 0.5$ ,  $\text{CH}_2\text{Cl}_2$ ); HPLC (AS column; eluent *i*PrOH/hexane 3:97; flow:  $v = 0.35$  mL/min;  $\lambda = 220$  nm): 15.9 min (1*S*,5*S*) and 19.7 min (1*R*,5*R*).

#### 4.17. Rearrangement of 3f

From *exo*-3f-*rac* and *endo*-3f-*rac* (200 mg, 0.62 mmol), lithium iodide (83 mg, 0.62 mmol). Yield: 80 mg (95%) of a mixture of *cis*-4f and *trans*-4f in a ratio 85:15.

#### 4.18. Rearrangement of *exo*-3g

From *exo*-3g (260 mg, 0.78 mmol), lithium iodide (104 mg, 0.78 mmol). Yield: 105 mg (90%) of a mixture of *cis*-4f and *trans*-4f in a ratio 85:15 (92% ee). HPLC (AS column; eluent *i*PrOH/hexane 3:97; flow:  $v = 1.00$  mL/min;  $\lambda = 220$  nm): 15.7 min (1*R*,7*S*), 17.3 min (1*R*,7*R*), 19.3 min (1*S*,7*S*) and 21.4 min (1*S*,7*R*).

**4.18.1. (1*R*,7*S*)-Bicyclo[6.3.0]dec-9-en-8-one (*cis*-4f).** RN 81255-92-7;  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.55 (dd,  $J = 5.6$ , 2.0 Hz, 1H), 6.16 (dd,  $J = 5.6$ , 2.0 Hz, 1H), 3.13 (m, 1H), 2.50 (dt,  $J = 10.5$ , 6.5 Hz, 1H), 2.02 (m, 1H), 1.92 (m, 1H), 2.10–1.20 (m, 8H);  $^{13}\text{C NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  212.7, 168.7, 132.3, 50.4, 47.4, 31.0, 30.6, 28.1, 27.5; M (GC-IE): 150 (100%), 107 (60%), 95 (48%).

**4.18.2. (1*R*,7*R*)-Bicyclo[6.3.0]dec-9-en-8-one (*trans*-4f).** RN 81255-93-8;  $^1\text{H NMR}$  (200 MHz,  $\text{CDCl}_3$ )  $\delta$  7.46 (d,  $J = 5.8$  Hz, 1H), 6.28 (d,  $J = 5.8$  Hz, 1H), 2.62 (d,  $J = 9.4$  Hz, 1H), 2.15–1.85 (m, 2H), 1.80–1.20 (m, 11H);  $^{13}\text{C NMR}$  (50 MHz,  $\text{CDCl}_3$ )  $\delta$  209.1 163.3, 134.8, 53.4, 47.4, 32.8, 31.1, 27.1, 27.0, 22.8; M (GC-IE): 150 (100%), 107 (60%), 95 (48%).

#### 4.19. Rearrangement of *exo*-3h

From *exo*-3h (167 mg, 0.48 mmol), lithium iodide (64 mg, 0.48 mmol). Yield: 68 mg (87%) of a mixture of *cis*-4h and *trans*-4h in a ratio 15:85 (98% ee). HPLC (AS column; eluent *i*PrOH/hexane 3:97; flow:  $v = 1.00$  mL/min;  $\lambda = 220$  nm): 12.5 min (1*R*,8*S*), 15.5 min (1*R*,8*R*), 21.4 min (1*S*,8*S*), and 26.0 min (1*S*,8*R*).

**4.19.1. (1*R*,8*S*)-Bicyclo[6.3.0]dec-9-en-8-one (*cis*-4h).**  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.53 (dd,  $J = 5.6$ , 2.8 Hz, 1H), 6.11 (dd,  $J = 5.6$ , 2.3 Hz, 1H), 2.94 (m, 1H), 2.15 (m, 1H), 2.10–1.20 (m, 8H); M (GC-IE): 164 (100%), 135 (68%).

**4.19.2. (1*R*,8*R*)-Bicyclo[6.3.0]dec-9-en-8-one (*trans*-4h).**  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$  7.44 (dd,  $J = 5.6$ , 2.2 Hz, 1H), 6.20 (ddd,  $J = 5.6$ , 2.2, 0.9 Hz, 1H), 2.72 (d,  $J = 12.5$  Hz, 1H), 2.15–1.85 (m, 3H), 1.85–1.65 (m, 4H), 1.60–1.30 (m, 4H), 1.20–1.05 (m, 2H);  $^{13}\text{C NMR}$  (75 MHz,  $\text{CDCl}_3$ )  $\delta$  212.3167.9, 131.6, 129.2, 51.5, 47.4, 34.1, 29.6, 27.3, 26.8, 25.3, 25.1; M (GC-IE): 164 (100%), 135 (36%), 120 (44%), 107 (48%), 94 (36%).

#### 4.20. Rearrangement of 3i

From *exo*-3i-*rac* and *endo*-3i-*rac* (150 mg, 0.44 mmol),

lithium iodide (60 mg, 0.44 mmol). Yield: 65 mg (94%) of 4i. 4-Phenylcyclopent-2-en-1-one (4i): RN 81255-96-1.

#### 4.21. Rearrangement of 3j

From *exo*-3j-*rac* and *endo*-3j-*rac* (166 mg, 0.5 mmol), lithium iodide (70 mg, 0.5 mmol). Yield: 68 mg (96%) of 4j. 4-Butylcyclopent-2-en-1-one (4j): RN 54814-22-1.

### Acknowledgments

This work was generously supported by the 'Ministère de l'Éducation et de la Recherche, Communauté française de Belgique' (action concertée, conventions 96/01-197) and the University of Louvain. We thank Professor E. de Hoffmann for the mass spectra.

### References

- Selected reviews: (a) Trost, B. M. *Top. Curr. Chem.* **1986**, *133*, 3. (b) Wong, H. N. C.; Lau, K. L.; Tam, K. F. *Top. Curr. Chem.* **1986**, *133*, 83. (c) Krief, A. *Top. Curr. Chem.* **1987**, *135*, 1. (d) Burger, P.; Buch, H. M. *Top. Curr. Chem.* **1987**, *135*, 77. (e) Bellus, D.; Ernst, B. *Ang. Chem. Int. Ed. Engl.* **1988**, *27*, 797. (f) Lee-Ruff, E. *Advances in Strain in Organic Chemistry*; Halton, B., Ed.; JAI: London, 1991; Vol. 1, pp 167–213.
- (a) Review: Ghosez, L.; Chen, L. Y.; Gobeaux, B.; Houge, C.; Markó, I. E.; Perry, M.; Saimoto, H. In *Strain and its Implications in Organic Chemistry*; de Meijere, A., Blechert, S., Eds.; Kluwer: Dordrecht, 1989; pp 235–254. (b) Marchand-Brynaert, J.; Ghosez, L. *J. Am. Chem. Soc.* **1972**, *72*, 2870–2872. (c) Falmagne, J. B.; Escudero, J. C.; Taleb-Sahraoui, S.; Ghosez, L. *Ang. Chem. Int. Ed. Engl.* **1981**, *20*, 879–880.
- (a) Greene, A. E.; Charbonnier, F. *Tetrahedron Lett.* **1985**, *26*, 5525–5529. (b) Greene, A. E.; Charbonnier, F.; Luche, M. J.; Moyano, A. *J. Am. Chem. Soc.* **1987**, *109*, 4752–4753. (c) Redlich, H.; Lenfers, J. B.; Kopf, J. *Ang. Chem. Int. Ed. Engl.* **1989**, *28*, 777–778. (d) Cagnon, J. R.; Le Bideau, F.; Marchand-Brynaert, J.; Ghosez, L. *Tetrahedron Lett.* **1997**, *38*, 2291–2294.
- (a) Houge, C.; Frisque-Hesbain, A. M.; Mockel, A.; Ghosez, L. *J. Am. Chem. Soc.* **1982**, *104*, 2920–2921. (b) Saimoto, H.; Houge, C.; Frisque, A. M.; Mockel, A.; Ghosez, L. *Tetrahedron Lett.* **1983**, *24*, 2251–2254. (c) Chen, L. Y.; Ghosez, L. *Tetrahedron Lett.* **1990**, *31*, 4467–4470. (d) Chen, L. Y.; Ghosez, L. *Tetrahedron Asym.* **1991**, *2*, 1181–1184. (e) Ghosez, L.; Genicot, C.; Gouverneur, V. *Pure Appl. Chem.* **1992**, *64*, 1849–1856. (f) Adam, J. M.; Ghosez, L.; Houk, K. N. *Ang. Chem. Int. Ed.* **1999**, *18*, 2728–2730.
- (a) Genicot, C.; Gobeaux, B.; Ghosez, L. *Tetrahedron Lett.* **1991**, *32*, 3827–3830. (b) Genicot, C.; Ghosez, L. *Tetrahedron Lett.* **1992**, *33*, 7357–7360. (c) Ghosez, L.; Mahuteau-Betzer, F.; Genicot, C.; Vallribera, A.; Cordier, J. F. *Chem. Eur. J.* **2002**, in press.
- Mahuteau-Betzer, F.; Ghosez, L. *Tetrahedron Lett.* **1999**, *40*, 5183–5186.
- See for instance: (a) Greene, A. E.; Deprès, J. P.; Crabbé, P. *Tetrahedron Lett.* **1977**, 2365–2368. (b) Mock, W. L.; Hartman, M. E. *J. Org. Chem.* **1977**, *42*, 466. (c) Greene,

- A. E.; Luche, M. J.; Deprès, J. P. *J. Am. Chem. Soc.* **1983**, *105*, 2435–2439. (d) Greene, A. E.; Luche, M. J.; Serra, A. A. *J. Org. Chem.* **1985**, *50*, 3957–3962. (e) Reeder, L. M.; Hegedus, L. S. *J. Org. Chem.* **1999**, *64*, 3306–3311.
8. (a) Leriverend, M. L.; Leriverend, P. C. R. *Acad. Sc., Ser. C* **1975**, *280*, 791–792. (b) Leriverend, M. L.; Leriverend, P. *Chem. Ber.* **1976**, *109*, 3492–3495. (c) Trost, B. M.; Latimer, L. H. *J. Org. Chem.* **1978**, *43*, 1031–1040. (d) Morton, D. R.; Brokaw, F. C. *J. Org. Chem.* **1979**, *44*, 2880–2887. (e) Krief, A.; Halazy, S. *J. Chem. Soc., Chem. Commun.* **1982**, 1200–1201. (f) Halazy, S.; Zutterman, F.; Krief, A. *Tetrahedron Lett.* **1982**, *23*, 4385–4388. (g) Tobe, Y.; Yamashita, S.; Yamashita, T.; Kakiuchi, K.; Odaira, Y. *J. Chem. Soc., Chem. Commun.* **1984**, 1259–1260. (h) Tobe, Y.; Yamashita, T.; Kakiuchi, K.; Odaira, Y. *J. Chem. Soc., Chem. Commun.* **1985**, 898–899. (i) Hart, T. W.; Comte, M. T. *Tetrahedron Lett.* **1985**, *26*, 2713–2716. (j) Pirrung, M. C.; Thomson, S. A. *J. Org. Chem.* **1988**, *53*, 227–230.
9. (a) Corey, E. J.; Chaykovsky, M. *J. Am. Chem. Soc.* **1962**, *84*, 3782–3783. (b) Review: Trost, B. M.; Melvin, JrL. S. *Sulfur Ylides: Emerging Synthetic Intermediates*; Academic: New York, 1975; pp 145–151.
10. Feneau-Dupont, J.; Declercq, J. P.; Vanwetswinckel, S.; Genicot, C. *Acta Crystallogr.* **1993**, *49*, 561–565.
11. Tinant, B.; Mahuteau-Betzer, F.; Ghosez, L. *Z. Kryst.* **2002**, Two manuscripts submitted for publication.
12. Rey, M.; Roberts, S. M.; Dreiding, A. S.; Roussel, A.; Vanlierde, H.; Toppet, S.; Ghosez, L. *Helv. Chim. Acta* **1982**, *65*, 703–720.
13. Compound **5b** was assigned a wrong structure in our preliminary account of this work.<sup>6</sup>